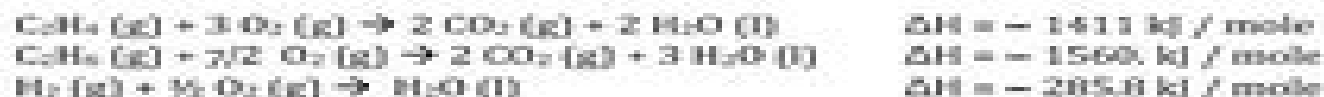


CHEM 12 Hess's Law Worksheet

1. Calculate the ΔH for the reaction: $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$,
from the following data:



2. Calculate the ΔH for the reaction:
 $4 NH_3(g) + 5 O_2(g) \rightarrow 4 NO(g) + 6 H_2O(g)$, from the following data:



3. Find the ΔH_f° (heat of formation) for acetic acid, $HC_2H_3O_2$, using the following thermochemical data:



Fun Fact: C(s) is in the form of graphite.

Acetic acid, $HC_2H_3O_2$, actually has the structure: CH_3COOH



4. Calculate the ΔH for the reaction: $CH_4(g) + NH_3(g) \rightarrow HCN(g) + 3 H_2(g)$,
from the following data:



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Solution and Electrolysis Sir William Cecil Dampier Dampier, 1895 Physical Chemistry Peter Atkins, Julio de Paula, Ronald Friedman, 2014 This title takes an innovative molecular approach to the teaching of physical chemistry The authors present the subject in a rigorous but accessible manner allowing students to gain a thorough understanding of physical chemistry **Prentice Hall Chemistry**, 2000 **Solution & Electrolysis** William Cecil Dampier Whetham, 1895

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