

## 19.2

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate  $K_a$  and  $K_b$  for weak acids and bases, and vice versa, given  $K_a$  or  $K_b$ .
2. Calculate  $K_a$  and  $K_b$  for weak acids and bases, and vice versa, given  $pK_a$  or  $pK_b$ .
3. Calculate the percent ionization of weak acids.

## Vocabulary

- |                            |                   |                             |
|----------------------------|-------------------|-----------------------------|
| acid dissociation constant | 1. conjugate acid | 2. weak acid                |
| acid ionization            | 3. base           | 4. base ionization constant |
|                            | 5. weak base      |                             |

## Key Equations

- |                                    |                                   |
|------------------------------------|-----------------------------------|
| 1. $K_a = \frac{[H^+][A^-]}{[HA]}$ | 2. $K_b = \frac{[HB][OH^-]}{[B]}$ |
| 3. $pK_a = -\log K_a$              | 4. $pK_b = -\log K_b$             |

## Part A Completion

Write the appropriate words in the blanks to complete the following statements. Write the chemical formula of the acid or base in the blank.

1. The  $K_a$  of acetic acid is  $1.8 \times 10^{-5}$ .
2. The  $K_b$  of ammonia is  $1.8 \times 10^{-5}$ .
3. The  $K_a$  of acetic acid is  $1.8 \times 10^{-5}$ .
4. The  $K_b$  of ammonia is  $1.8 \times 10^{-5}$ .
5. The  $K_a$  of acetic acid is  $1.8 \times 10^{-5}$ .
6. The  $K_b$  of ammonia is  $1.8 \times 10^{-5}$ .
7. The  $K_a$  of acetic acid is  $1.8 \times 10^{-5}$ .
8. The  $K_b$  of ammonia is  $1.8 \times 10^{-5}$ .
9. The  $K_a$  of acetic acid is  $1.8 \times 10^{-5}$ .
10. The  $K_b$  of ammonia is  $1.8 \times 10^{-5}$ .

1. acetic acid
2.  $1.8 \times 10^{-5}$
3.  $1.8 \times 10^{-5}$
4.  $1.8 \times 10^{-5}$
5.  $1.8 \times 10^{-5}$
6.  $1.8 \times 10^{-5}$
7.  $1.8 \times 10^{-5}$
8.  $1.8 \times 10^{-5}$
9.  $1.8 \times 10^{-5}$
10.  $1.8 \times 10^{-5}$

# 19 2 Section Review Hydrogen Ions And Acidity

**Zhe Xu**

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the author's name.

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