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(1975-1995)
VAUXHALL



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standardization of sodium hydroxide naoh and oxalic acid $\text{H}_2\text{C}_2\text{O}_4$ $2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ COOK aq $\text{C}_6\text{H}_4\text{COONa}$ COOK aq H_2O the point of equivalence is the point of titration when the number of

moles of naoh na added is equal to the number of moles of khp nb in the