

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of a given acid or hydrogen ion concentration.
2. Calculate hydrogen ion concentrations from pH and pOH , and vice versa from pH and pOH .
3. Calculate the pH of a weak acid.

Vocabulary

- | | | |
|-------------------|-------------------------------|------------------|
| 1. autoionization | 2. hydrogen ion concentration | 3. acid solution |
| 4. autoionization | 5. pH | 6. pOH |

Key Equation

- | | |
|-------------------------------------|----------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = 14 - \text{pOH}$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = 14 - \text{pOH}$ |

Part A Completion

Write the hydrogen ion concentration or autoionization equilibrium of the acids and bases that are indicated in the spaces provided. Be sure to include the correct state symbols in the chemical equations.

1. What substance is _____, a weak hydrogen ion concentration of $1.0 \times 10^{-4} \text{ M}$ is the concentration of this acidic substance in water and equal to _____ M .
2. Hydrochloric acid ionizes from _____ to H^+ and _____.
3. _____, a concentrated, strong acid, ionizes completely _____.
4. _____, a strong acid, ionizes _____.
5. _____, a weak acid, has a pH of _____.
6. _____, a weak acid, has a pH of _____.
7. _____, a strong acid, has a pH of _____.

8. _____
9. _____
10. _____
11. _____
12. _____
13. _____
14. _____
15. _____
16. _____
17. _____
18. _____

19 2 Hydrogen Ions And Acidity Work Sheet Answers

J Spring



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Class 10th Science Worksheet , This book is as per the guidelines syllabus and marking scheme issued by CBSE for Class X The salient features of this workbook are The questions in the this book have been so designed that complete syllabus is covered This book help students to identify their weak areas and improve them Additional it will help students gain confidence The questions in the book are of varying difficulty level and will help students evaluate their reasoning analysis and understanding of the subject matter **Topics-aids** Rod O'Connor,1975 **Maple® for Environmental Sciences**

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