

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions of use?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, what factor determines whether a reaction occurs?
- ☒ ΔG
☐ ΔH
☐ ΔS
☐ ΔG°

3. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☒ 25°C and 100°C

4. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, $\Delta G^\circ = -237 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1 mole of H_2O at standard conditions?
- ☐ 1 mole
☐ 0.5 mole
☒ 2 moles
☐ 4 moles

5. Water is the most abundant molecule in the atmosphere.
- ☐ 100%
☒ 100%
☐ 100%
☐ 100%

6. Water is the most abundant molecule in the atmosphere.
- ☐ 100%
☐ 100%
☒ 100%
☐ 100%

7. A reaction's standard state is 100°C at 1 atm. What is the standard state for water?
- ☒ 100°C
☐ 100°C
☐ 100°C
☐ 100°C

8. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, $\Delta G^\circ = -237 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1 mole of H_2O at standard conditions?
- ☒ 1 mole
☐ 1 mole
☐ 1 mole
☐ 1 mole

9. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, $\Delta G^\circ = -237 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1 mole of H_2O at standard conditions?
- ☐ 1 mole
☐ 1 mole
☒ 1 mole
☐ 1 mole

10. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☐ 25°C and 100°C

11. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☐ 25°C and 100°C

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