

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?  
 a. 1  
 b. 10  
 c. 100  
 d. 1000

2. At temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2$  has kinetic energy  $E_1$ . At temperature  $T_2$ , what kinetic energy does  $1 \text{ mol}$  of  $\text{H}_2$  have?  
 a.  $E_1$   
 b.  $2E_1$   
 c.  $4E_1$   
 d.  $8E_1$

3. The universal gas constant is the same for:  
 a. all of the above  
 b. gases with different molar masses  
 c. liquid water  
 d. solid substances

4. For temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2$  has kinetic energy  $E_1$ . For temperature  $T_2$ ,  $1 \text{ mol}$  of  $\text{H}_2$  has kinetic energy  $E_2$ . What is the ratio of  $E_2$  to  $E_1$ ?  
 a.  $E_2/E_1$   
 b.  $2E_2/E_1$   
 c.  $4E_2/E_1$   
 d.  $8E_2/E_1$

5. What is the study of the relationship between mechanical work and heat called?  
 a. classical thermodynamics  
 b. statistical thermodynamics  
 c. physical chemistry  
 d. chemical thermodynamics

6. What is the basic natural distribution?  
 a. Boltzmann  
 b. Maxwell-Boltzmann  
 c. Fermi-Dirac  
 d. Bose-Einstein

7. A substance's empirical formula is  $\text{CH}_2$ . At the same time, it is known to be a gas at 100°C and 1 atm. What is its molecular formula?  
 a.  $\text{CH}_2$   
 b.  $\text{C}_2\text{H}_4$   
 c.  $\text{C}_3\text{H}_6$   
 d.  $\text{C}_4\text{H}_8$

8. For temperature  $T_1$ ,  $1 \text{ mol}$  of  $\text{H}_2$  has kinetic energy  $E_1$ . For temperature  $T_2$ ,  $1 \text{ mol}$  of  $\text{H}_2$  has kinetic energy  $E_2$ . What is the ratio of  $E_2$  to  $E_1$ ?  
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 a. all of the above  
 b. gases with different molar masses  
 c. liquid water  
 d. solid substances

# Multiple Choice Stoichiometry Test

**John T. Moore, Richard H. Langley**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located on the right side of the slide.

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## Reviewing **Multiple Choice Stoichiometry Test**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Multiple Choice Stoichiometry Test**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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