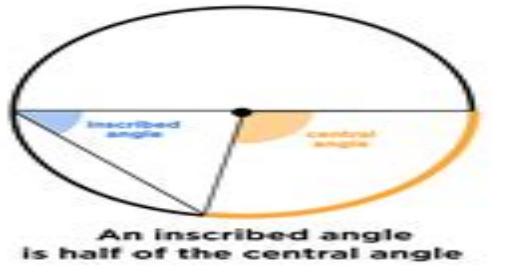
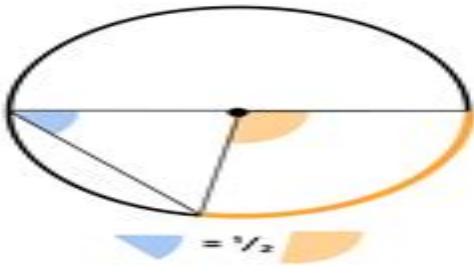
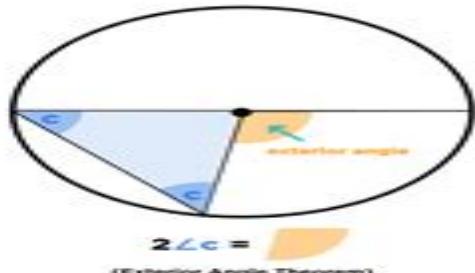
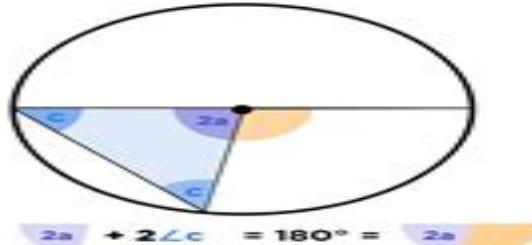
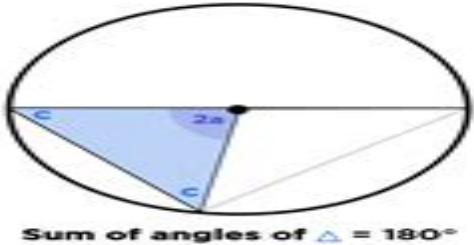
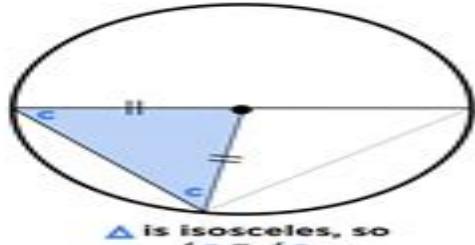
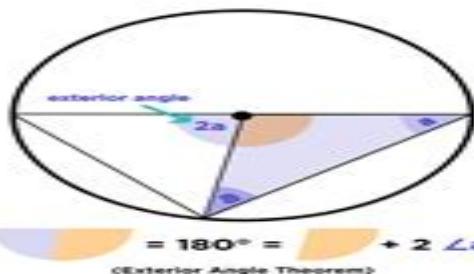
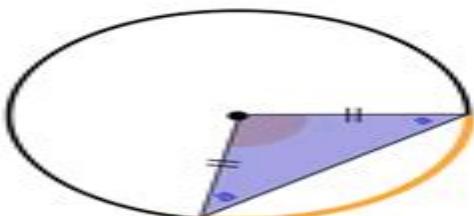
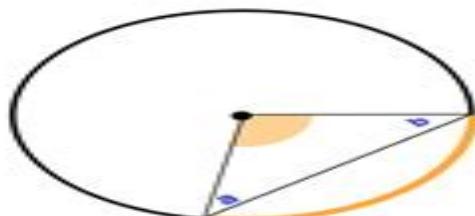
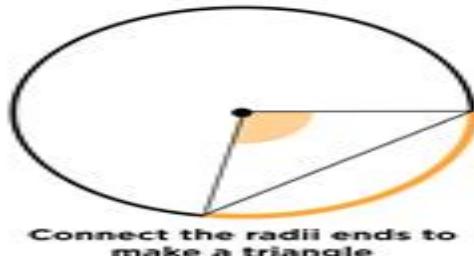
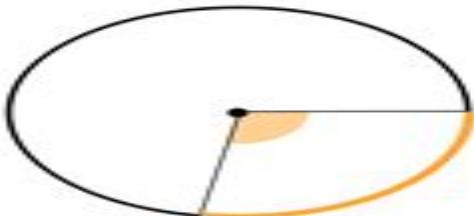
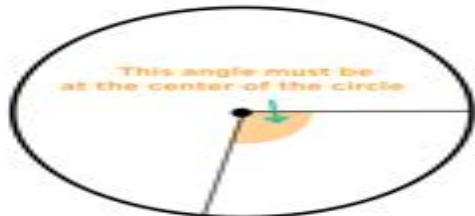


## Inscribed Angle Theorem

Special Case: Inscribed angle includes a diameter



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web if you don t know how to calculate the number of protons neutrons electrons or nucleons that are in an atom of an element reading this page may help you would you like to practice only the first 36 elements would you like the challenge of

having questions with just big atomic numbers

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web the easiest way to find the atomic number is to look on a periodic table the atomic number is in the upper left corner or is the largest number on the square finding the number of protons the number of protons in an atom is equal to the atomic number of the element for example let s use oxygen

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web jun 2 2019 number of protons atomic number of the element number of electrons number of protons number of neutrons mass number atomic number find the number of protons each element is defined by the number of protons found in each of its atoms no matter how many electrons or neutrons an atom has the element is defined

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