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## ACS CHEMISTRY FINAL EXAM STUDY GUIDE

- milli (m) (m has 3 lines, upside down it's a w (win) so negative=loss):  $10^{-3}$
- micro (inappropriate joke, i don't want Quizlet to ban me lol):  $10^{-6}$
- nano:  $10^{-9}$
- pico (p):  $10^{-12}$
- mono: one, single, alone
- di: two
- tri: three (tricycle)
- tetra: four (tetrapod) - dogs, cats, vertebrae animals
- penta: five (pentagon)
- hexa: 6 (hexagon)
  
- hepta: 7 (heptagon)
- octa: 8 (octopus, stop)
- nona: 9 (no = neign in German)
- deca: 10 (decade = 10 years)
- molar mass: the mass of one mole of a pure substance
- molar ratio: ratio of coefficients in a chemical equation
- limiting reactant: A reactant that is totally consumed during a chemical reaction, limits the extent of the reaction, and determines the amount of product.
- percent composition: the percent by mass of each element in a compound
- molarity: the number of moles of solute per liter of solution
- percent yield:  $\text{actual yield} / \text{theoretical yield} \times 100$
- empirical formula: a chemical formula showing the ratio of elements in a compound rather than the total number of atoms
- molecular formula: A chemical formula that shows the number and kinds of atoms in a molecule, but not the arrangement of the atoms.
- gas laws: describe how the behavior of gases is affected by pressure, volume, and temperature

# 2nd Semester Chemistry Final Exam Study Guide

**Illinois State Board of Health, John  
Henry Rauch**



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## **Table of Contents 2nd Semester Chemistry Final Exam Study Guide**

1. Understanding the eBook 2nd Semester Chemistry Final Exam Study Guide
  - The Rise of Digital Reading 2nd Semester Chemistry Final Exam Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying 2nd Semester Chemistry Final Exam Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 2nd Semester Chemistry Final Exam Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from 2nd Semester Chemistry Final Exam Study Guide
  - Personalized Recommendations
  - 2nd Semester Chemistry Final Exam Study Guide User Reviews and Ratings
  - 2nd Semester Chemistry Final Exam Study Guide and Bestseller Lists
5. Accessing 2nd Semester Chemistry Final Exam Study Guide Free and Paid eBooks
  - 2nd Semester Chemistry Final Exam Study Guide Public Domain eBooks
  - 2nd Semester Chemistry Final Exam Study Guide eBook Subscription Services
  - 2nd Semester Chemistry Final Exam Study Guide Budget-Friendly Options

6. Navigating 2nd Semester Chemistry Final Exam Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - 2nd Semester Chemistry Final Exam Study Guide Compatibility with Devices
  - 2nd Semester Chemistry Final Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 2nd Semester Chemistry Final Exam Study Guide
  - Highlighting and Note-Taking 2nd Semester Chemistry Final Exam Study Guide
  - Interactive Elements 2nd Semester Chemistry Final Exam Study Guide
8. Staying Engaged with 2nd Semester Chemistry Final Exam Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 2nd Semester Chemistry Final Exam Study Guide
9. Balancing eBooks and Physical Books 2nd Semester Chemistry Final Exam Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 2nd Semester Chemistry Final Exam Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 2nd Semester Chemistry Final Exam Study Guide
  - Setting Reading Goals 2nd Semester Chemistry Final Exam Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 2nd Semester Chemistry Final Exam Study Guide
  - Fact-Checking eBook Content of 2nd Semester Chemistry Final Exam Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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