



**SECTION – B**  
**(8 marks each)**

Answer **any five** questions in **not more than 2 pages each** :

**(8×5=40)**

2. Briefly explain the sources of Indian mercantile law.
3. Define and distinguish between coercion and undue influence.
4. Who is a minor ? Explain the liabilities of minor for necessities.
5. Explain the duties of an agent.
6. Explain the case "Carlill Vs. Carbolic smoke ball Co."
7. Define a bill of exchange. Distinguish between a cheque and bill of exchange.
8. Briefly explain the types of settlement (arbitration, mediation and conciliation) under Arbitration Act.

**SECTION – C**  
**(20 marks each)**

Answer the following in **not more than 6 pages each** :

**(20×3=60)**

9. Define and explain the term offer with its essentials.

**OR**

Define a contract. Explain the essentials of a valid contract.

10. Explain the various agreements opposed to public policy.

**OR**

Explain the remedies available for breach of contract.

11. Who is an agent ? What are the general principles of agency ? Explain the different modes of creation of agency.

**OR**

Define and explain the contract of guarantee. Explain the rights of surety.

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web jan 23 2023 formation the formation of radicals may involve breaking of covalent bonds homolytically a process that requires significant amounts of energy for example splitting  $\text{H}_2$  into  $2\text{H}$  has a  $\Delta h$  of 435 kJ/mol and  $\text{Cl}_2$  into  $2\text{Cl}$  has a  $\Delta h$  of 243 kJ/mol this is known as the homolytic bond dissociation energy and is usually abbreviated as the

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web oct 6 2021 answer 5 3 2 notice that the terms involving the square root in the denominator are eliminated by multiplying by the conjugate we can use the property  $a/b \cdot a/b = a^2/b^2$  to expedite the process of multiplying the expressions in the denominator example 5 4 16 rationalize the denominator 10 2 6

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web in chemistry a radical also known as a free radical is an atom molecule or ion that has at least one unpaired valence electron 1 2 with some exceptions these unpaired electrons make radicals highly chemically reactive many radicals spontaneously dimerize most organic radicals have short lifetimes