

## Carbonate Ores Method

### Hypothesis:

"The amount of carbonate in an ore sample can be measured by adding hydrochloric acid to the sample until it stops fizzing."

### Equipment:

- 100ml beaker
- Samples of different carbonate ores labeled 10%, 15%, 20%, 25%, 30% of copper carbonate.
- Measuring cylinder
- Stopwatch
- 2M Hydrochloric acid
- Spatula

### Method:

1. Put two spatulas of an ore sample into the beaker.
2. Measure 25ml of 2M hydrochloric acid using the measuring cylinder.
3. Pour the dilute hydrochloric acid into the beaker and start the stopwatch. Swirl the beaker once.
4. Measure the time taken for the fizzing to stop.
5. Repeat for the other samples.

### Variables:

**Independent Variable:** Percentage of Copper Carbonate in the ore (%)

**Dependent Variable:** Time taken for the fizzing to stop (s)

**Control Variables:** Volume of acid, concentration of acid.

### Risk Assessment:

**Hazard:** Hydrochloric acid

**Risk:** Could get into your eyes and cause irritation.

**Control:** Wear safety glasses at all times when doing the investigation

**Hazard:** Copper Carbonate

**Risk:** It is a harmful chemical and could get into your eyes or on your skin.

**Control:** Wear safety glasses at all times when doing the investigation. If you get it on your skin, wash off with cold water.

### Title:

Carbonate Ores Method

### Web address:

<http://monocytes.co.uk/teaching-resources>

# Method Of Carbonate Ores Isa Practical Aqa

**William Mack,William Benjamin Hale**

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## **Table of Contents Method Of Carbonate Ores Isa Practical Aqa**

1. Understanding the eBook Method Of Carbonate Ores Isa Practical Aqa
  - The Rise of Digital Reading Method Of Carbonate Ores Isa Practical Aqa
  - Advantages of eBooks Over Traditional Books
2. Identifying Method Of Carbonate Ores Isa Practical Aqa
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Method Of Carbonate Ores Isa Practical Aqa
  - User-Friendly Interface
4. Exploring eBook Recommendations from Method Of Carbonate Ores Isa Practical Aqa

- Personalized Recommendations
- Method Of Carbonate Ores Isa Practical Aqa User Reviews and Ratings
- Method Of Carbonate Ores Isa Practical Aqa and Bestseller Lists

5. Accessing Method Of Carbonate Ores Isa Practical Aqa Free and Paid eBooks

- Method Of Carbonate Ores Isa Practical Aqa Public Domain eBooks
- Method Of Carbonate Ores Isa Practical Aqa eBook Subscription Services
- Method Of Carbonate Ores Isa Practical Aqa Budget-Friendly Options

6. Navigating Method Of Carbonate Ores Isa Practical Aqa eBook Formats

- ePUB, PDF, MOBI, and More
- Method Of Carbonate Ores Isa Practical Aqa Compatibility with Devices
- Method Of Carbonate Ores Isa Practical Aqa Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Method Of Carbonate Ores Isa Practical Aqa
- Highlighting and Note-Taking Method Of Carbonate Ores Isa Practical Aqa
- Interactive Elements Method Of Carbonate Ores Isa Practical Aqa

8. Staying Engaged with Method Of Carbonate Ores Isa Practical Aqa

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Method Of Carbonate Ores Isa Practical Aqa

9. Balancing eBooks and Physical Books Method Of Carbonate Ores Isa Practical Aqa

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Method Of Carbonate Ores Isa Practical Aqa

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Method Of Carbonate Ores Isa Practical Aqa

- Setting Reading Goals Method Of Carbonate Ores Isa Practical Aqa
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Method Of Carbonate Ores Isa Practical Aqa

- Fact-Checking eBook Content of Method Of Carbonate Ores Isa Practical Aqa
- Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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