

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate K_a and K_b for weak acids and bases, and calculate pK_a and pK_b for weak acids and bases.
2. Calculate K_a and K_b for weak acids and bases, and calculate pK_a and pK_b for weak acids and bases.
3. Calculate the pH of weak acid and base solutions.

Vocabulary

- | | | |
|-----------------|--------------------------------------|------------------------------------|
| acid | acid dissociation constant (K_a) | acid ionization constant (K_a) |
| acidic solution | acidic solution | acidic solution |
| acidic solution | acidic solution | acidic solution |

Key Equations

- | | |
|---|--|
| $K_a = \frac{[H^+][A^-]}{[HA]}$ (Equation 19.1) | $K_b = \frac{[OH^-][HB^+]}{[B]}$ (Equation 19.2) |
|---|--|

Part A Completion

Write the chemical formula for each of the following acids, bases, and salts. Identify the acid, base, or salt.

1. Hydrochloric acid, HCl , (Equation 19.1)
2. Acetic acid, CH_3COOH , (Equation 19.1)
3. Nitric acid, HNO_3 , (Equation 19.1)
4. Sulfuric acid, H_2SO_4 , (Equation 19.1)
5. Hydrofluoric acid, HF , (Equation 19.1)
6. Hydrobromic acid, HBr , (Equation 19.1)
7. Hydroiodic acid, HI , (Equation 19.1)
8. Hydrocyanic acid, HCN , (Equation 19.1)
9. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
10. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
11. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
12. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
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17. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
18. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)
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20. Hydroperfluoroboric acid, $HFBO_2$, (Equation 19.1)

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Maurice Bleifeld



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