

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000?
 a. 1
 b. 1000
 c. 100
 d. 10

2. At temperature T_1 , 1 mol of H_2 has kinetic energy E_1 . At temperature T_2 , 1 mol of H_2 has kinetic energy E_2 .
 a. $E_2 = E_1$
 b. $E_2 = 2E_1$
 c. $E_2 = 4E_1$
 d. $E_2 = 8E_1$

3. The equilibrium shift down in the forward rxn.
 a. Shift of the reaction
 b. Shift of the reaction
 c. Shift of the reaction
 d. Shift of the reaction

4. For the reaction $\text{H}_2 + \text{Cl}_2 \rightleftharpoons 2 \text{HCl}$, $K_c = 4.0$. If the initial concentration of H_2 is 1.0 M and the initial concentration of Cl_2 is 1.0 M , what is the equilibrium concentration of HCl ?
 a. 1.0 M
 b. 2.0 M
 c. 4.0 M
 d. 8.0 M

5. What is the study of the relationship between the chemical and physical properties of substances?
 a. Chemical thermodynamics
 b. Chemical thermodynamics
 c. Chemical thermodynamics
 d. Chemical thermodynamics

6. What is the kinetic energy of a molecule?
 a. $\frac{1}{2}mv^2$
 b. $\frac{1}{2}mv^2$
 c. $\frac{1}{2}mv^2$
 d. $\frac{1}{2}mv^2$

7. A substance's empirical formula is CH_2 . At the same time, it is also known that the substance is a gas at 100°C and 1 atm . What is the molecular formula?
 a. CH_2
 b. CH_4
 c. C_2H_4
 d. C_3H_6

8. For the reaction $\text{H}_2 + \text{Cl}_2 \rightleftharpoons 2 \text{HCl}$, $K_c = 4.0$. If the initial concentration of H_2 is 1.0 M and the initial concentration of Cl_2 is 1.0 M , what is the equilibrium concentration of HCl ?
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 c. $\frac{1}{2}mv^2$
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 a. Shift of the reaction
 b. Shift of the reaction
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 d. Shift of the reaction

Multiple Choice Stoichiometry Test

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