

| Ques                       | Ans | Questions   | Marking |
|----------------------------|-----|---|---------|
| Ques 1<br>Ques 2<br>Ques 3 | 8.  | (a) Describe the technique of measuring the rate of reaction between hydrogen peroxide and potassium iodide in the presence of a catalyst.                        | 5       |
|                            | 12  | (a) Write down the balanced chemical equation for the reaction between hydrogen peroxide and potassium iodide in the presence of a catalyst.                      | 4       |
|                            | A   | (c) A gas occupies 100 ml volume at 10°C and 1 atm pressure. Calculate the volume of this gas at 20°C and 1 atm pressure if the temperature is increased 4 times. | 4       |
| Ques 1<br>Ques 2<br>Ques 3 | 8.  | <b>UNIT IV - 1998-99</b><br>(a) Describe the properties of Newton's corpuscular theory along with its short comings.  | 7       |
|                            | 12  | (b) Describe the properties of Newton's corpuscular theory along with its short comings.  | 3       |
|                            | A   | (c) Write about the following with respect to the properties of light:<br>i. Reflection<br>ii. Refraction<br>iii. Dispersion                                      | 4       |

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