

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At standard T_p , 1 mol of H_2O vapor has higher enthalpy than liquid water.
 a. True
 b. False
 c. Not sure
 d. None

3. The enthalpy of fusion is the amount of:
 a. Heat of the system
 b. Heat of the surroundings
 c. Heat of the universe
 d. Heat of the system and surroundings

4. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, the heat of reaction is -286 kJ/mol . How much heat is released when 1 mole of H_2 reacts with 1 mole of O_2 ?
 a. 286 kJ
 b. 572 kJ
 c. 143 kJ
 d. 286 kJ

5. What is the study of the relationship between mechanical work and heat called?
 a. Thermodynamics
 b. Kinetics
 c. Equilibrium
 d. None of the above

6. What is the heat of fusion of water?
 a. 334 J/mol
 b. 334 kJ/mol
 c. 334 J/g
 d. 334 kJ/g

7. A substance's specific heat is $1 \text{ J/g}^\circ\text{C}$. At this rate, how much heat is needed to raise the temperature of 1 kg of the substance by 1°C ?
 a. 1 J
 b. 1 kJ
 c. 1000 J
 d. 1000 kJ

8. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, the heat of reaction is -286 kJ/mol . How much heat is released when 1 mole of H_2 reacts with 1 mole of O_2 ?
 a. 286 kJ
 b. 572 kJ
 c. 143 kJ
 d. 286 kJ

9. At standard T_p , 1 mol of H_2O vapor has higher enthalpy than liquid water.
 a. True
 b. False
 c. Not sure
 d. None

10. The enthalpy of fusion of water is 334 J/g . How much heat is needed to melt 1 kg of ice at 0°C ?
 a. 334 J
 b. 334 kJ
 c. 334 J/g
 d. 334 kJ/g

11. The enthalpy of fusion of water is 334 J/g . How much heat is needed to melt 1 kg of ice at 0°C ?

Multiple Choice Stoichiometry Test

N Colangelo



Multiple Choice Stoichiometry Test:

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