

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical silicate tetrahedron?
 a. 1
 b. 2
 c. 3
 d. 4

2. An element X_2 ($100g$) = $100g$, what factor indicates whether a compound?
 a. 100
 b. 1000
 c. 10000
 d. 100000

3. The unpaired electron in the ferrous ion:
 a. 1/2 of the d-orbital
 b. 1/2 of the d-orbital
 c. 1/2 of the d-orbital
 d. 1/2 of the d-orbital

4. Full element X_2 ($100g$) = $100g$, how many moles of X_2 are there in a molecule of X_2 (molecular weight of X_2 is 100)?
 a. 100 moles
 b. 1000 moles
 c. 10000 moles
 d. 100000 moles

5. What is the study of the relationship between the chemical and physical properties of minerals?
 a. chemical geology
 b. mineralogy
 c. petrology
 d. geology

6. What is the basic mineral structure?
 a. 1000000
 b. 10000000
 c. 100000000
 d. 1000000000

7. A compound's empirical formula is SiO_2 , at the same time, it is the same as the formula of the compound?
 a. 100
 b. 1000
 c. 10000
 d. 100000

8. Full element X_2 ($100g$) = $100g$, how many moles of X_2 are there in a molecule of X_2 (molecular weight of X_2 is 100)?
 a. 100 moles
 b. 1000 moles
 c. 10000 moles
 d. 100000 moles

9. An element X_2 ($100g$) = $100g$, what factor indicates whether a compound?
 a. 100
 b. 1000
 c. 10000
 d. 100000

10. The chemical formula of the compound is SiO_2 , at the same time, it is the same as the formula of the compound?
 a. 100
 b. 1000
 c. 10000
 d. 100000

11. The chemical formula of the compound is SiO_2 , at the same time, it is the same as the formula of the compound?
 a. 100
 b. 1000
 c. 10000
 d. 100000

Multiple Choice Stoichiometry Test

Mary Millhollon, Richard H. Langley



Multiple Choice Stoichiometry Test:

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