

Semester 2 Final Exam Review – Answer Key

1. Mendeleev
2. Electrons in the outer-most energy shell of an atom
3. F, O, N
4. -ide
5. Atoms with a full valence shell
6. Molarity (M) = moles per liter = moles/liters
7. Basic, alkaline
8. Group or Family
9. $6.02 \times 10^{23} \times 5 = 3.01 \times 10^{24}$
10. Fluid, compressible, take the shape and volume of the container, molecules are far apart (low density)
11. Period
12. Acidic
13. High boiling point, low vapor pressure
14. A homogeneous mixture of only one phase; solute dissolved in a solvent
15. 22.4L
16. Endothermic
17. $\text{Zn}(\text{NO}_3)_2$
18. H, N, O, F, Cl, Br, I
19. Nature of the reactants, temperature, surface area, concentration, catalyst
20. Calorie = energy required to raise 1g of water 1 degree Celcius
21. Volume is directly proportional to temperature @ a constant pressure ($V_1/T_1 = V_2/T_2$)
22. Iron(II) chloride
23. $\text{Ba}(\text{NO}_3)_2$
24. PbCO_3
25. Reactants react to form products
26. Synthesis, decomposition, combustion, single replacement, double replacement
27. Water and carbon dioxide
28. Right side, $-\Delta H$
29. The energy require to form an activated complex and have a reaction proceed to form products
30. Exothermic, $\Delta H = -585\text{kJ}$
 - Convert mass to moles: $34\text{g} + 17\text{g/mol} = 2 \text{ mol NH}_3$
 - Multiply by energy/mole ratio: $2 \text{ mol NH}_3 \times (-1170\text{kJ})/4 \text{ mol NH}_3 = -585\text{kJ}$
31. Concentration, temperature, pressure (gas, heterogeneous system)
32. A buffer solution is a solution of a conjugate acid-base pair. A buffer system prevents significant changes in pH when a small amount of acid or base is added to the solution. Your body has a bicarbonate buffer system in the blood which helps maintain you blood's pH.
33. Unsaturated = more room for solute @ temp; saturated = solute maxed out @ temp; supersaturated = more solute than maximum for given temp

Nd Semester Final Exam Review Science Answer

Joacim Rocklöv



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web the three year plan includes guidelines and proposals in several policy priority areas e g improve the quality of teachers school leadership and schools through self evaluations update the curriculum ensure all day school provision and develop further policy actions addressing the specific education levels of early childhood education to