

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions of use?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, what factor determines whether a reaction occurs?
- ☒ ΔG
☐ ΔH
☐ ΔS
☐ ΔG°

3. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☒ 25°C and 100°C

4. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, $\Delta G^\circ = -237 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1 mole of H_2O at standard conditions?
- ☐ 1 mole
☐ 0.5 mole
☒ 2 moles
☐ 4 moles

5. What is the state of the water molecule in the liquid state?
- ☐ discrete molecules
☒ hydrogen bonding
☐ crystal lattice
☐ ionic interactions

6. What is the boiling point of water?
- ☐ 0°C
☐ 100°C
☒ 212°F
☐ 32°F

7. A substance's empirical formula is CH_2 . At the same time, it is known to be a gas at 1 atm and 100°C. What is the molecular formula?
- ☒ C_2H_4
☐ C_3H_6
☐ C_4H_8
☐ C_5H_{10}

8. For the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, $\Delta G^\circ = -237 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1 mole of H_2O at standard conditions?
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10. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☐ 25°C and 100°C

11. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm
☐ 25°C and 1 atm
☒ 1 atm and 100°C
☐ 25°C and 100°C

Multiple Choice Stoichiometry Test

Sasha Alcott



Multiple Choice Stoichiometry Test:

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