

Multiple Choice

Select the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at 100°C and 100 atm?  
a. 1  
b. 2  
c. 3  
d. 4
2. At constant  $T_1 = 100^\circ\text{C}$ , what factor determines whether a process is spontaneous?  
a.  $\Delta H$   
b.  $\Delta G$   
c.  $\Delta S$   
d.  $\Delta T$
3. The standard state of water is the liquid state.  
a. Not at any pressure  
b. pure water/ice/water  
c. pure water/steam  
d. pure water/ice
4. For the reaction  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ , the free energy change is  $-237 \text{ kJ/mol}$  at 25°C. What is the standard Gibbs free energy of formation of water at 25°C?  
a. 237 kJ/mol  
b. -237 kJ/mol  
c. 474 kJ/mol  
d. -474 kJ/mol
5. What is the state of the water in the following reaction?  $\text{H}_2\text{O}(l) \rightarrow \text{H}_2\text{O}(g)$   
a. liquid water  
b. gaseous water  
c. solid water  
d. liquid water
6. What is the boiling point of water?  
a. 0°C  
b. 100°C  
c. 273°C  
d. 373°C
7. A substance's standard state is 100°C, at this state, it is in the liquid state. What is the standard state of water?  
a. 100°C  
b. 0°C  
c. 273°C  
d. 373°C
8. For the reaction  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ , the free energy change is  $-237 \text{ kJ/mol}$  at 25°C. What is the standard Gibbs free energy of formation of water at 25°C?  
a. 237 kJ/mol  
b. -237 kJ/mol  
c. 474 kJ/mol  
d. -474 kJ/mol
9. The standard state of water is the liquid state.  
a. Not at any pressure  
b. pure water/ice/water  
c. pure water/steam  
d. pure water/ice
10. The standard state of water is the liquid state.  
a. Not at any pressure  
b. pure water/ice/water  
c. pure water/steam  
d. pure water/ice

# Multiple Choice Stoichiometry Test

**DP Hallahan**



## Multiple Choice Stoichiometry Test:

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