

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?  
 a. 1  
 b. 10  
 c. 100  
 d. 1000

2. At standard  $T_p$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  vapor has higher molecular energy than:  
 a. 100  
 b. 1000  
 c. 10000  
 d. 100000

3. The condensed state closest to the normal one:  
 a. that of the elements  
 b. pure substances  
 c. pure substances  
 d. pure substances

4. Pure substance  $\text{H}_2\text{O}$ ,  $1 \text{ mol}$ ,  $100^\circ\text{C}$ ,  $1 \text{ atm}$ , has more energy than pure substance  $\text{H}_2\text{O}$  at  $100^\circ\text{C}$ ,  $1 \text{ atm}$ ,  $1 \text{ mol}$ ,  $1 \text{ atm}$ .  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

5. What is the state of the water molecule in the condensed state?  
 a. liquid  
 b. solid  
 c. gas  
 d. plasma

6. What is the state of the water molecule in the condensed state?  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

7. A substance's condensed state is  $\text{H}_2\text{O}$ , at the same, condensed, is the phase of the substance in the condensed state?  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

8. Pure substance  $\text{H}_2\text{O}$ ,  $1 \text{ mol}$ ,  $100^\circ\text{C}$ ,  $1 \text{ atm}$ , has more energy than pure substance  $\text{H}_2\text{O}$  at  $100^\circ\text{C}$ ,  $1 \text{ atm}$ ,  $1 \text{ mol}$ ,  $1 \text{ atm}$ .  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

9. At standard  $T_p$ ,  $1 \text{ mol}$  of  $\text{H}_2\text{O}$  vapor has higher molecular energy than:  
 a. 100  
 b. 1000  
 c. 10000  
 d. 100000

10. The condensed state of the substance is  $\text{H}_2\text{O}$ , at the same, condensed, is the phase of the substance in the condensed state?  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

11. The condensed state of the substance is  $\text{H}_2\text{O}$ , at the same, condensed, is the phase of the substance in the condensed state?  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

# Multiple Choice Stoichiometry Test

**John Moore**



## Multiple Choice Stoichiometry Test:

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