

#### References and Bibliography

Because of the high concentration of the product, the reaction mixture was extremely viscous. The reaction mixture with the addition of the starting ether dissolved in a solution of  $\text{LiClO}_4$  and  $\text{CH}_3\text{CN}$  was then deoxygenated. After a 10-min. period, the reaction mixture was heated to 100°C. for 1 hr. The reaction mixture was then diluted with ether and the product was isolated by vacuum distillation. The yield of the product was 70% of the theoretical amount. The product was a colorless, crystalline, solid, m.p. 100–102°C.

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