

## 19.2 Hydrogen Ions and Acidity

- The reaction in which water molecules produce ions is called the self-ionization of water
- $\text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- $2\text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$



# 19 2 Hydrogen Ions And Acidity

**C. Neal, C. J. Smith, S. Hill**



## 19 2 Hydrogen Ions And Acidity:

Carbonic Acid Compounds and Hydrogen Ion Activities in Blood and Salt Solutions Erik Johan Warburg, 1922

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