

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on K_a and K_b and vice versa.
3. Calculate the pH of weak acids and bases.

Vocabulary

- | | | |
|------------------------|-----------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base |
| 4. ionization | 5. K_a | 6. K_b |
| 7. ionization constant | 8. pH | |

Key Equation

- | | |
|-----------------------|-----------------------|
| 1. $[H^+] = 10^{-pH}$ | 2. $pH = -\log [H^+]$ |
| 3. $pH = -\log [H^+]$ | 4. $[H^+] = 10^{-pH}$ |

Part A Completion

Write the equation and/or the numerical value(s) of the constant(s) that are indicated in the spaces provided. Be sure to include correct units when appropriate.

1. Water autoionizes. K_w is the ion-product constant for water. Write the equation and K_w at 25°C.
2. The autoionization of water is endothermic. Write the equation and K_w at 50°C.
3. The autoionization of water is endothermic. Write the equation and K_w at 0°C.
4. The autoionization of water is endothermic. Write the equation and K_w at 100°C.
5. The autoionization of water is endothermic. Write the equation and K_w at 150°C.
6. The autoionization of water is endothermic. Write the equation and K_w at 200°C.
7. The autoionization of water is endothermic. Write the equation and K_w at 250°C.
8. The autoionization of water is endothermic. Write the equation and K_w at 300°C.
9. The autoionization of water is endothermic. Write the equation and K_w at 350°C.
10. The autoionization of water is endothermic. Write the equation and K_w at 400°C.
11. The autoionization of water is endothermic. Write the equation and K_w at 450°C.
12. The autoionization of water is endothermic. Write the equation and K_w at 500°C.
13. The autoionization of water is endothermic. Write the equation and K_w at 550°C.
14. The autoionization of water is endothermic. Write the equation and K_w at 600°C.
15. The autoionization of water is endothermic. Write the equation and K_w at 650°C.
16. The autoionization of water is endothermic. Write the equation and K_w at 700°C.
17. The autoionization of water is endothermic. Write the equation and K_w at 750°C.
18. The autoionization of water is endothermic. Write the equation and K_w at 800°C.
19. The autoionization of water is endothermic. Write the equation and K_w at 850°C.
20. The autoionization of water is endothermic. Write the equation and K_w at 900°C.
21. The autoionization of water is endothermic. Write the equation and K_w at 950°C.
22. The autoionization of water is endothermic. Write the equation and K_w at 1000°C.

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19 2 Hydrogen Ions And Acidity Worksheet Answers

Sir Tarun Rupani



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Table of Contents 19 2 Hydrogen Ions And Acidity Worksheet Answers

1. Understanding the eBook 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - The Rise of Digital Reading 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Personalized Recommendations
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers User Reviews and Ratings
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers and Bestseller Lists

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 - 19 2 Hydrogen Ions And Acidity Worksheet Answers eBook Subscription Services
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Budget-Friendly Options
6. Navigating 19 2 Hydrogen Ions And Acidity Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Compatibility with Devices
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Highlighting and Note-Taking 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Interactive Elements 19 2 Hydrogen Ions And Acidity Worksheet Answers
8. Staying Engaged with 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 19 2 Hydrogen Ions And Acidity Worksheet Answers
9. Balancing eBooks and Physical Books 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 19 2 Hydrogen Ions And Acidity Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Setting Reading Goals 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Fact-Checking eBook Content of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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