

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At standard T_p , 1 mol of H_2O vapor has higher molecular energy than:
 a. 100
 b. 1000
 c. 10000
 d. 100000

3. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

4. First Bornemann H_2 , 1 mol of H_2 gas is at 100°C and 1 atm . How many moles of H_2 gas are at 100°C and 1 atm ?
 a. 100
 b. 1000
 c. 10000
 d. 100000

5. What is the state of the water molecules in the condensed state?
 a. liquid
 b. solid
 c. gas
 d. plasma

6. What is the state of the water molecules in the condensed state?
 a. 100000
 b. 10000
 c. 1000
 d. 100

7. A sample of water at 100°C at 1 atm is at the normal boiling point. How many moles of water are in the sample?
 a. 100
 b. 1000
 c. 10000
 d. 100000

8. First Bornemann H_2 , 1 mol of H_2 gas is at 100°C and 1 atm . How many moles of H_2 gas are at 100°C and 1 atm ?
 a. 100
 b. 1000
 c. 10000
 d. 100000

9. At standard T_p , 1 mol of H_2O vapor has higher molecular energy than:
 a. 100
 b. 1000
 c. 10000
 d. 100000

10. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

11. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

Multiple Choice Stoichiometry Test

L Reisser



Multiple Choice Stoichiometry Test:

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