

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm and 27°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$ and $T_2 = 200^\circ\text{C}$, what factor multiplies the average kinetic energy?
- ☒ 2
☐ 3
☐ 4
☐ 5

3. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both of these
☐ none of these

4. For a constant $T_1 = 100^\circ\text{C}$ and $T_2 = 200^\circ\text{C}$, how many times is the average kinetic energy of a molecule increased at T_2 as at T_1 (approx)?
- ☐ 2
☐ 3
☐ 4
☒ 5

5. What is the study of the interrelationships among environmental problems in natural systems?
- ☐ systems engineering
☒ systems engineering
☐ systems engineering
☐ systems engineering

6. What is the basic unit of measurement?
- ☐ 1.000000
☐ 1.000000
☒ 1.000000
☐ 1.000000

7. A constant's average kinetic energy is 100°C at the same distance as the gas is 100°C at the same distance. The factor is:
- ☒ 100
☐ 100
☐ 100
☐ 100

8. For a constant $T_1 = 100^\circ\text{C}$ and $T_2 = 200^\circ\text{C}$, how many times is the average kinetic energy of a molecule increased at T_2 as at T_1 (approx)?
- ☒ 2
☐ 3
☐ 4
☐ 5

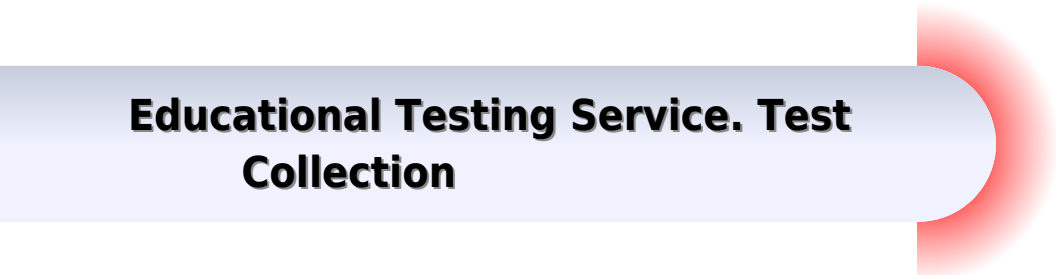
9. At constant $T_1 = 100^\circ\text{C}$ and $T_2 = 200^\circ\text{C}$, what factor multiplies the average kinetic energy?
- ☐ 2
☐ 3
☒ 4
☐ 5

10. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both of these
☐ none of these

11. The universal gas constant is the same as:
- ☐ Boltzmann's constant
☐ gas constant
☒ both of these
☐ none of these

Multiple Choice Stoichiometry Test

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