

CHEM 12 Hess's Law Worksheet

1. Calculate the ΔH for the reaction: $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$,
from the following data:



2. Calculate the ΔH for the reaction:
 $4 NH_3(g) + 5 O_2(g) \rightarrow 4 NO(g) + 6 H_2O(g)$, from the following data:



3. Find the ΔH_f° (heat of formation) for acetic acid, $HC_2H_3O_2$, using the following thermochemical data:



Fun Fact: C(s) is in the form of graphite.

Acetic acid, $HC_2H_3O_2$, actually has the structure: CH_3COOH



4. Calculate the ΔH for the reaction: $CH_4(g) + NH_3(g) \rightarrow HCN(g) + 3 H_2(g)$,
from the following data:



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DJ Losen



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Thermodynamics Ricardo Morales-Rodriguez, 2012-10-03 This book presents the selection of various high level contributions involving thermodynamics The book goes from the fundamentals up to several applications in different scientific fields The content of the book has been classified in six sections Classical Thermodynamics Statistical Thermodynamics Property Prediction in Thermodynamics Material and Products Non Equilibrium and Thermodynamics in Diverse Areas The classification of the book aims to provide to the reader the facility of finding the desired topic included in the book It is expected that this collection of chapters will contribute to the state of the art in the thermodynamics area

Elements of Physical Chemistry Peter William Atkins, Julio De Paula, 2017 This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal particularly to students with an interest in biological applications

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