

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many atoms of hydrogen (protons) is a molecule of liquid water made of?  
 a. 1  
 b. 2  
 c. 3  
 d. 4

2. At standard  $H_2O$  (1 atm) = 100°C, what factor determines whether a substance:  
 a. Boils  
 b. Freezes  
 c. Melts  
 d. Condenses

3. The condensed state closest to the normal one:  
 a. Most of the elements  
 b. Most of the molecules  
 c. Most of the atoms  
 d. Most of the particles

4. Pure substance  $H_2O$  (1 atm) = 0°C, 100°C, how many atoms of oxygen (protons) are present in a molecule of water at all conditions?  
 a. 10 atoms  
 b. 100 atoms  
 c. 1000 atoms  
 d. 10000 atoms

5. What is the state of the water molecules in a solid state?  
 a. Disordered  
 b. Ordered  
 c. Random  
 d. None of the above

6. What is the boiling point of water?  
 a. 0°C  
 b. 100°C  
 c. 212°F  
 d. 373K

7. A substance's condensed state is  $H_2O$ , at the same distance to the ground state, the substance is:  
 a. Solid  
 b. Liquid  
 c. Gas  
 d. Plasma

8. Pure substance  $H_2O$  (1 atm) = 100°C, how many (protons of a molecule) of oxygen (protons) are present in a molecule?  
 a. 1000  
 b. 10000  
 c. 100000  
 d. 1000000

9. At standard  $H_2O$  (1 atm) = 100°C, 1 atm, what factor determines whether a substance will:  
 a. Boil  
 b. Freeze  
 c. Melt  
 d. Condense

10. What is the state of the water molecules in a solid state?  
 a. Disordered  
 b. Ordered  
 c. Random  
 d. None of the above

11. The boiling point of water is 100°C, how many atoms of oxygen (protons) are present in a molecule?

# Multiple Choice Stoichiometry Test

**Tao Wei**



## Multiple Choice Stoichiometry Test:

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