

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions of use?
- ☐ 1  
☒ 2  
☐ 3  
☐ 4

2. For the reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ , what factor determines whether a reaction occurs?
- ☒  $\Delta G$   
☐  $\Delta H$   
☐  $\Delta S$   
☐  $\Delta G^\circ$

3. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm  
☐ 25°C and 1 atm  
☒ 0°C and 1 atm  
☐ 100°C and 0.1 atm

4. For the reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ,  $\Delta G^\circ = -237 \text{ kJ/mol}$ . How many moles of  $\text{H}_2$  are required to produce 1 mole of  $\text{H}_2\text{O}$  at standard conditions?
- ☐ 1 mole  
☐ 0.5 mole  
☒ 2 moles  
☐ 4 moles

5. Water is the most abundant molecule in the atmosphere.
- ☐ 100%  
☒ 100%  
☐ 100%  
☐ 100%

6. Water is the most abundant molecule in the atmosphere.
- ☐ 100%  
☐ 100%  
☒ 100%  
☐ 100%

7. A reaction's standard state is 100°C at 1 atm, standard is 100°C at 1 atm, standard is 100°C at 1 atm, standard is 100°C at 1 atm.
- ☐ 100°C  
☒ 100°C  
☐ 100°C  
☐ 100°C

8. For the reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ,  $\Delta G^\circ = -237 \text{ kJ/mol}$ . How many moles of  $\text{H}_2$  are required to produce 1 mole of  $\text{H}_2\text{O}$  at standard conditions?
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☐ 0.5 mole  
☐ 2 moles  
☐ 4 moles

9. For the reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ,  $\Delta G^\circ = -237 \text{ kJ/mol}$ . How many moles of  $\text{H}_2$  are required to produce 1 mole of  $\text{H}_2\text{O}$  at standard conditions?
- ☐ 1 mole  
☐ 0.5 mole  
☒ 2 moles  
☐ 4 moles

10. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm  
☒ 25°C and 1 atm  
☐ 0°C and 1 atm  
☐ 100°C and 0.1 atm

11. The standard state of water is the liquid state.
- ☐ 100°C and 1 atm  
☒ 25°C and 1 atm  
☐ 0°C and 1 atm  
☐ 100°C and 0.1 atm

# Multiple Choice Stoichiometry Test

**John T. Moore, Richard H. Langley**



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