

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At temperature T_1 , 1 mol of H_2O vapor has kinetic energy E_1 . At temperature T_2 , 1 mol of H_2O vapor has kinetic energy E_2 .
 a. $E_2 = E_1$
 b. $E_2 = 2E_1$
 c. $E_2 = 4E_1$
 d. $E_2 = 8E_1$

3. The enthalpy of fusion is the amount of:
 a. heat of the system
 b. heat of the surroundings
 c. heat of the system
 d. heat of the surroundings

4. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, $\Delta H^\circ = -285.8 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1000 kJ of heat?
 a. 1000 mol
 b. 100 mol
 c. 10 mol
 d. 1 mol

5. What is the study of the relationship between the structure and properties of materials?
 a. chemical engineering
 b. materials science
 c. physics
 d. chemistry

6. What is the basic unit of measurement for energy?
 a. 1 Joule
 b. 100 Joules
 c. 1000 Joules
 d. 10000 Joules

7. A substance's specific heat is $1 \text{ J/g}^\circ\text{C}$. At the same time, it is $100 \text{ J/g}^\circ\text{C}$. How much heat is required to raise the temperature of 1 g of the substance by 1°C ?
 a. 100 J
 b. 1 J
 c. 1000 J
 d. 10000 J

8. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, $\Delta H^\circ = -285.8 \text{ kJ/mol}$. How many moles of H_2 are required to produce 1000 kJ of heat?
 a. 1000 mol
 b. 100 mol
 c. 10 mol
 d. 1 mol

9. At the same time, it is $100 \text{ J/g}^\circ\text{C}$. At the same time, it is $1000 \text{ J/g}^\circ\text{C}$. How much heat is required to raise the temperature of 1 g of the substance by 1°C ?
 a. 100 J
 b. 1 J
 c. 1000 J
 d. 10000 J

10. The enthalpy of fusion is the amount of:
 a. heat of the system
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11. The enthalpy of fusion is the amount of:
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 b. heat of the surroundings
 c. heat of the system
 d. heat of the surroundings

Multiple Choice Stoichiometry Test

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