

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical soapwater, OH^- ?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, what factor determines whether a reaction?
- ☒ 1
☐ 2
☐ 3
☐ 4

3. The equilibrium shift occurs in the forward direction.
- ☐ 1. Shift of the reaction
☐ 2. Shift of the reaction
☒ 3. Shift of the reaction
☐ 4. Shift of the reaction

4. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, what factor determines whether a reaction is spontaneous?
- ☐ 1. Shift of the reaction
☐ 2. Shift of the reaction
☒ 3. Shift of the reaction
☐ 4. Shift of the reaction

5. What is the study of the relationship between the chemical and physical properties of a substance?
- ☐ 1. Chemical equilibrium
☒ 2. Chemical equilibrium
☐ 3. Chemical equilibrium
☐ 4. Chemical equilibrium

6. What is the study of the relationship between the chemical and physical properties of a substance?
- ☐ 1. Chemical equilibrium
☐ 2. Chemical equilibrium
☒ 3. Chemical equilibrium
☐ 4. Chemical equilibrium

7. A substance's chemical formula is H_2O , at the same time, it is also known as the product of the reaction.
- ☐ 1. H_2O
☒ 2. H_2O
☐ 3. H_2O
☐ 4. H_2O

8. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, what factor determines whether a reaction is spontaneous?
- ☒ 1. Shift of the reaction
☐ 2. Shift of the reaction
☐ 3. Shift of the reaction
☐ 4. Shift of the reaction

9. For the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$, what factor determines whether a reaction is spontaneous?
- ☐ 1. Shift of the reaction
☐ 2. Shift of the reaction
☒ 3. Shift of the reaction
☐ 4. Shift of the reaction

10. What is the study of the relationship between the chemical and physical properties of a substance?
- ☐ 1. Chemical equilibrium
☒ 2. Chemical equilibrium
☐ 3. Chemical equilibrium
☐ 4. Chemical equilibrium

11. The equilibrium shift occurs in the forward direction.

Multiple Choice Stoichiometry Test

John T. Moore, Richard H. Langley



Multiple Choice Stoichiometry Test:

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