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	METHOD STATEMENT	Issue Date	12 September 2013
	Method statement for the blinding concrete (PCC) works	Rev. Date	NA
		Pages	1 of 8

METHOD STATEMENT FOR THE BLINDING CONCRETE (PCC) WORKS

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3 655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 000731 moles making the experimental molar mass

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 \text{aq}$ $\text{NaOH \text{aq}}$ $\text{Na}_2\text{C}_2\text{O}_4 \text{4H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction $\text{NaOH \text{aq}}$ $\text{C}_6\text{H}_4\text{COOH \text{aq}}$ $\text{C}_6\text{H}_4\text{COONa \text{aq}}$ $\text{H}_2\text{O \text{l}}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP added in the