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Safeway Customer Service Training Manual

JG Myers

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3 655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the

number of moles of the acid in the solution to be 000731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the k_a or k_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate k_{a1} and k_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1 hcl aq naoh aq nacl aq h₂o 1 hcl a q naoh a q nacl a q h 2 o l one mole of hcl hcl would be fully neutralized by one mole of naoh naoh

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used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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