

ACS Final Exam Practice 1

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is $3.0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$. What is this speed in $\text{nm}\cdot\text{ms}^{-1}$?

- (A) $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$ (B) $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$ (C) $3.0 \times 10^{14} \text{ nm}\cdot\text{ms}^{-1}$ (D) $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

- (A) 1.3 g (B) 1.30 g (C) 1.298 g (D) 1.2980 g

PQ-5. What is the name of $\text{Ti}_2(\text{PO}_4)_3$?

- (A) titanium phosphate (B) titanium(III) phosphate
(C) titanium(IV) phosphate (D) titanium tetraphosphate

PQ-6. The formula of strontium hexafluorosilicate is SrSiF_6 . What is the formula of aluminum hexafluorosilicate?

- (A) AlSiF_6 (B) $\text{Al}_2(\text{SiF}_6)_3$ (C) Al_2SiF_6 (D) $\text{Al}_2(\text{SiF}_6)_2$

PQ-8. What is the correct name for N_2O_5 ?

- (A) nitrogen(II) oxide (B) nitrogen(V) oxide
(C) dinitrogen oxide (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of $12.0 \text{ g}\cdot\text{cm}^{-3}$, what is the mass of a nanoparticle with a volume of $1.84 \times 10^{-23} \text{ cm}^3$?

- (A) $2.21 \times 10^{-23} \text{ g}$ (B) $1.63 \times 10^{-20} \text{ g}$ (C) $6.52 \times 10^{-21} \text{ g}$ (D) $1.53 \times 10^{-22} \text{ g}$

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Encyclopedia of Chemical Processing and Design John J. McKetta Jr,1993-06-02 Written by engineers for engineers with over 150 International Editorial Advisory Board members this highly lauded resource provides up to the minute information on the chemical processes methods practices products and standards in the chemical and related industries

Lessons in Organic Chemistry G. S. Turpin,1894 Inorganic Chemistry for Beginners Henry Enfield Roscoe,Joseph Lunt,1897 **The Publishers Weekly** ,1892 *Science Reporter* ,1994 *Boston Journal of Chemistry and Popular Science Review* ,1881 **Report Upon the Survey of the University of Wisconsin** Wisconsin. State Board of Public Affairs,1914 **A Class-book of Inorganic Chemistry; with tables of chemical analysis, etc** David MORRIS (B.A.),1870

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Table of Contents 70 Question Chemistry Final Exam Answers

1. Understanding the eBook 70 Question Chemistry Final Exam Answers
 - The Rise of Digital Reading 70 Question Chemistry Final Exam Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 70 Question Chemistry Final Exam Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 70 Question Chemistry Final Exam Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 70 Question Chemistry Final Exam Answers
 - Personalized Recommendations
 - 70 Question Chemistry Final Exam Answers User Reviews and Ratings
 - 70 Question Chemistry Final Exam Answers and Bestseller Lists
5. Accessing 70 Question Chemistry Final Exam Answers Free and Paid eBooks

- 70 Question Chemistry Final Exam Answers Public Domain eBooks
 - 70 Question Chemistry Final Exam Answers eBook Subscription Services
 - 70 Question Chemistry Final Exam Answers Budget-Friendly Options
6. Navigating 70 Question Chemistry Final Exam Answers eBook Formats
- ePub, PDF, MOBI, and More
 - 70 Question Chemistry Final Exam Answers Compatibility with Devices
 - 70 Question Chemistry Final Exam Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 70 Question Chemistry Final Exam Answers
 - Highlighting and Note-Taking 70 Question Chemistry Final Exam Answers
 - Interactive Elements 70 Question Chemistry Final Exam Answers
8. Staying Engaged with 70 Question Chemistry Final Exam Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 70 Question Chemistry Final Exam Answers
9. Balancing eBooks and Physical Books 70 Question Chemistry Final Exam Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 70 Question Chemistry Final Exam Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 70 Question Chemistry Final Exam Answers
- Setting Reading Goals 70 Question Chemistry Final Exam Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 70 Question Chemistry Final Exam Answers
- Fact-Checking eBook Content of 70 Question Chemistry Final Exam Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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