

Multiple Choice

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions, 1000 L?
 a. 1
 b. 10
 c. 100
 d. 1000

2. At standard T_p , 1 mol of H_2O vapor has higher molecular energy than:
 a. 100
 b. 1000
 c. 10000
 d. 100000

3. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

4. First Bornemann H_2 , 1 mol of H_2 gas at 100°C has more energy than 1 mol of H_2 gas at 0°C .
 a. 100 J
 b. 1000 J
 c. 10000 J
 d. 100000 J

5. What is the state of the water molecules in the condensed state?
 a. random
 b. ordered
 c. partially ordered
 d. completely ordered

6. What is the boiling point of water?
 a. 100°C
 b. 100°F
 c. 212°F
 d. 373°C

7. A substance's condensed state is H_2O at the triple point. It is H_2O at the triple point. It is H_2O at the triple point.
 a. 100°C
 b. 100°F
 c. 212°F
 d. 373°C

8. First Bornemann H_2 , 1 mol of H_2 gas at 100°C has more energy than 1 mol of H_2 gas at 0°C .
 a. 100 J
 b. 1000 J
 c. 10000 J
 d. 100000 J

9. At standard T_p , 1 mol of H_2O vapor has higher molecular energy than:
 a. 100
 b. 1000
 c. 10000
 d. 100000

10. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

11. The condensed state closest to the normal one:
 a. that of the metals
 b. that of the nonmetals
 c. that of the metalloids
 d. that of the noble gases

Multiple Choice Stoichiometry Test

Thomas H. Martinson



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