

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of hydrogen gas are in a cylinder at 100 atm pressure, 20°C?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. At constant $T_1 = 100^\circ\text{C}$, what is the molar volume of hydrogen gas?
- ☒ 20
☐ 10
☐ 5
☐ 1

3. The molar volume of hydrogen is the same as:
- ☐ that of any gas
☐ that of any liquid
☒ that of hydrogen
☒ that of oxygen

4. For hydrogen H_2 , $T = 100^\circ\text{C}$, $p = 100\text{ atm}$, how many moles of H_2 are in a cylinder of 100 L volume?
- ☐ 10 moles
☐ 100 moles
☐ 1000 moles
☒ 10000 moles

5. What is the state of the gas in the cylinder?
- ☐ liquid
☒ solid
☒ gas
☐ liquid

6. What is the molar volume of hydrogen?
- ☐ 100 L
☐ 1000 L
☐ 10000 L
☒ 10 L

7. A cylinder of hydrogen gas is 100 L at 100 atm. How many moles of H_2 are in the cylinder?
- ☒ 100
☐ 1000
☐ 10000
☐ 100000

8. For hydrogen H_2 , $T = 100^\circ\text{C}$, $p = 100\text{ atm}$, how many moles of H_2 are in a cylinder of 100 L volume?
- ☒ 1000
☐ 100
☐ 10
☐ 1

9. At constant $T = 100^\circ\text{C}$, $p = 100\text{ atm}$, what is the molar volume of hydrogen gas?
- ☐ 10
☐ 100
☒ 1000
☐ 10000

10. What is the molar volume of hydrogen?
- ☐ 100 L
☒ 1000 L
☐ 10000 L
☐ 100000 L

11. The molar volume of H_2 is 10 L at 100 atm. How many moles of H_2 are in a cylinder of 100 L volume?

Multiple Choice Stoichiometry Test

Mark Steyvers



Multiple Choice Stoichiometry Test:

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