

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydrogen gas are in a cylinder at typical conditions of use?
- ☐ 1  
☒ 2  
☐ 3  
☐ 4

2. At standard  $T_p$  ( $T_p = 298K$ ), what factor determines whether a compound is a gas?
- ☒  $\Delta H_{vap}$   
☐  $\Delta H_{sub}$   
☐  $\Delta H_{fus}$   
☐  $\Delta H_{cond}$

3. The condensed state closest to the normal one:
- ☐ that of the elements  
☐ pure substances  
☒ pure substances  
☒ pure substances

4. For substance  $H_2$  ( $T_p = 298K$ ),  $\Delta H_{vap} = 9.1 kJ/mol$ . How many moles of  $H_2$  are vaporized by providing 100 kJ of heat at  $T_p$ ?
- ☐ 10 mol  
☐ 0.01 mol  
☐ 100 mol  
☒ 10 mol

5. What is the state of the water-vaporizing system at standard conditions of standard?
- ☐ liquid water  
☐ gaseous water  
☒ liquid water  
☒ liquid water

6. What is the boiling point of water?
- ☐ 0.01°C  
☐ 0.01°C  
☐ 0.01°C  
☒ 0.01°C

7. A substance's condensed state is  $H_2O$  at the point where it is given a large amount of heat.
- ☐  $H_2O$   
☒  $H_2O$   
☐  $H_2O$   
☐  $H_2O$

8. For substance  $H_2$  ( $T_p = 298K$ ),  $\Delta H_{vap} = 9.1 kJ/mol$ . How many moles of  $H_2$  are vaporized by providing 100 kJ of heat at  $T_p$ ?
- ☒ 10 mol  
☐ 0.01 mol  
☐ 100 mol  
☐ 10 mol

9. At standard  $T_p$  ( $T_p = 298K$ ),  $\Delta H_{vap} = 9.1 kJ/mol$ . How many moles of  $H_2$  are vaporized by providing 100 kJ of heat at  $T_p$ ?
- ☐ 10 mol  
☐ 0.01 mol  
☐ 100 mol  
☒ 10 mol

10. What is the boiling point of water?
- ☐ 0.01°C  
☒ 0.01°C  
☐ 0.01°C  
☐ 0.01°C

11. The boiling point of  $H_2O$  is 100°C. How many moles of  $H_2O$  are vaporized by providing 100 kJ of heat at  $T_p$ ?

# Multiple Choice Stoichiometry Test

**Richard C. Sweetland, Daniel J. Keyser**



## Multiple Choice Stoichiometry Test:

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