

# Chemistry Study Guide

Periodic Table of the Elements

The periodic table displays elements organized by atomic number, with groups labeled 1 through 10 and periods labeled 1 through 7. Lanthanide and actinide series are shown below the main table.

## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎📌

## Matter:

Anything that has **mass** and **occupies space** ✎🌐

Examples: air, gases, liquids, solids!

## Atoms:

Atoms are the **smallest, indivisible particle** ✎👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# Section 2 nuclear Chemistry Study Guide Key

**Donna Batton**



## **Section 2nuclear Chemistry Study Guide Key:**

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