

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many moles of hydroxide groups are a molecule of typical silicate tetrahedron?
 a. 1
 b. 2
 c. 3
 d. 4

2. For a mineral $\text{Si}_2\text{O}_5(\text{OH})_2$, what is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

3. The uncharged silicate anion in the feldspar ion:
 a. SiO_4^{4-}
 b. SiO_3^{2-}
 c. SiO_2
 d. SiO_4^{2-}

4. For a mineral $\text{Si}_2\text{O}_5(\text{OH})_2$, what is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

5. What is the study of the relationships among mineral products in mineral assemblages?
 a. mineralogy
 b. petrology
 c. mineralogy
 d. mineralogy

6. What is the basic mineral structure?
 a. 1:1
 b. 2:1
 c. 3:1
 d. 4:1

7. A mineral's empirical formula is $\text{Si}_2\text{O}_5(\text{OH})_2$, what is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

8. For a mineral $\text{Si}_2\text{O}_5(\text{OH})_2$, what is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

9. For a mineral $\text{Si}_2\text{O}_5(\text{OH})_2$, what is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

10. What is the charge balance coefficient?
 a. 0.5
 b. 1.0
 c. 1.5
 d. 2.0

11. The charge balance coefficient is 0.5 for a mineral with the empirical formula $\text{Si}_2\text{O}_5(\text{OH})_2$.

Multiple Choice Stoichiometry Test

John T. Moore, Richard H. Langley



Multiple Choice Stoichiometry Test:

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