

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common chemical gas used?

1. How many atoms of fluorine are present in a molecule of hydrofluoric acid?
- ☐ 1
☒ 2
☐ 3
☐ 4

2. An element X , $100g \rightarrow 100g$, when heated undergoes a chemical reaction.
- ☒ 100
☐ 100
☐ 100
☐ 100

3. The compound CH_4 is the chemical formula for:
- ☐ Methane
☐ Ethane
☒ Propane
☐ Butane

4. The compound CH_4 is the chemical formula for:
- ☐ Methane
☐ Ethane
☒ Propane
☐ Butane

5. What is the name of the compound CH_4 ?
- ☐ Methane
☒ Ethane
☐ Propane
☐ Butane

6. What is the name of the compound CH_4 ?
- ☐ Methane
☒ Ethane
☐ Propane
☐ Butane

7. A compound's empirical formula is CH_4 . At the same time, it is also known that the compound is a gas at room temperature. What is the molecular formula?
- ☒ CH_4
☐ CH_8
☐ CH_2
☐ CH_6

8. The compound CH_4 is the chemical formula for:
- ☒ Methane
☐ Ethane
☐ Propane
☐ Butane

9. An element X , $100g \rightarrow 100g$, when heated undergoes a chemical reaction.
- ☐ 100
☐ 100
☒ 100
☐ 100

10. The compound CH_4 is the chemical formula for:
- ☐ Methane
☒ Ethane
☐ Propane
☐ Butane

11. The compound CH_4 is the chemical formula for:
- ☐ Methane
☒ Ethane
☐ Propane
☐ Butane

Multiple Choice Stoichiometry Test

Sasha Alcott



Multiple Choice Stoichiometry Test:

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