

Name:

2021

Name:

Unit 3: Reading Section: 10 Marks

Section 1: 10-15: Writing & Mathematics Chemical Equations

1. Reading Section:

- Read the text and answer the questions in the spaces provided. (10 marks)
- Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water. (2 marks)
- Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form hydrogen peroxide. (2 marks)
- Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)
- Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

Section 2: 16-20: Writing & Mathematics Chemical Equations

2. Section 2: Writing & Mathematics Chemical Equations

16. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water. (2 marks)

17. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form hydrogen peroxide. (2 marks)

3. Section 3: Writing & Mathematics Chemical Equations

18. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

19. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

4. Section 4: Writing & Mathematics Chemical Equations



20. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

21. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

22. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

23. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

24. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

25. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

26. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

27. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

28. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

29. Write a balanced chemical equation for the reaction between hydrogen gas and oxygen gas to form water and hydrogen peroxide. (2 marks)

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