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Estimating the effect of the three variables on the likelihood of adopting the practices. Table 2 (not shown) presents the results.

1. **Q** 10. If the energy content of chloroplasts increases a substance is produced, what substance is this?  
 a. **A** ☐  $\text{H}_2$   
 b. **B** ☒  $\text{O}_2$   
 c. **C** ☐  $\text{H}_2\text{O}$   
 d. **D** ☐  $\text{CO}_2$
2. **Q** 10. At thermodynamic  $T_1 = 300\text{K}$ , a heat engine produces a thermodynamic work of  $W_1 = 100\text{J}$ .  
 a. **A** ☐  $100\text{J}$   
 b. **B** ☒  $100\text{J}$   
 c. **C** ☐  $100\text{J}$   
 d. **D** ☐  $100\text{J}$
3. **Q** 10. The equilibrium shift towards the forward reaction:  
 a. **A** ☐ shift of the forward reaction  
 b. **B** ☐ shift of the backward reaction  
 c. **C** ☐ shift of the forward reaction  
 d. **D** ☒ shift of the backward reaction
4. **Q** 10. First thermodynamic  $T_1 = 300\text{K}$ ,  $T_2 = 400\text{K}$ , heat engine produces a thermodynamic work of  $W_1 = 100\text{J}$ .  
 a. **A** ☐  $100\text{J}$   
 b. **B** ☒  $100\text{J}$   
 c. **C** ☐  $100\text{J}$   
 d. **D** ☐  $100\text{J}$
5. **Q** 10. When is the shift of the equilibrium towards the forward reaction?  
 a. **A** ☐ decrease of the forward reaction  
 b. **B** ☒ increase of the forward reaction  
 c. **C** ☐ decrease of the backward reaction  
 d. **D** ☐ increase of the backward reaction
6. **Q** 10. When is the shift of the equilibrium towards the forward reaction?  
 a. **A** ☐ decrease of the forward reaction  
 b. **B** ☒ increase of the forward reaction  
 c. **C** ☐ decrease of the backward reaction  
 d. **D** ☐ increase of the backward reaction
7. **Q** 10. A substance's equilibrium constant is  $K_{eq} = 100$ , at this temperature, the reaction is:  
 a. **A** ☐  $100$   
 b. **B** ☒  $100$   
 c. **C** ☐  $100$   
 d. **D** ☐  $100$
8. **Q** 10. First thermodynamic  $T_1 = 300\text{K}$ ,  $T_2 = 400\text{K}$ , heat engine produces a thermodynamic work of  $W_1 = 100\text{J}$ .  
 a. **A** ☐  $100\text{J}$   
 b. **B** ☒  $100\text{J}$   
 c. **C** ☐  $100\text{J}$   
 d. **D** ☐  $100\text{J}$
9. **Q** 10. At thermodynamic  $T_1 = 300\text{K}$ ,  $T_2 = 400\text{K}$ , heat engine produces a thermodynamic work of  $W_1 = 100\text{J}$ .  
 a. **A** ☐  $100\text{J}$   
 b. **B** ☒  $100\text{J}$   
 c. **C** ☐  $100\text{J}$   
 d. **D** ☐  $100\text{J}$
10. **Q** 10. The equilibrium constant for the reaction  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$  is  $K_{eq} = 100$ .  
 a. **A** ☐  $100$   
 b. **B** ☒  $100$   
 c. **C** ☐  $100$   
 d. **D** ☐  $100$
11. **Q** 10. The equilibrium constant for the reaction  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$  is  $K_{eq} = 100$ .  
 a. **A** ☐  $100$   
 b. **B** ☒  $100$   
 c. **C** ☐  $100$   
 d. **D** ☐  $100$
12. **Q** 10. The equilibrium constant for the reaction  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$  is  $K_{eq} = 100$ .  
 a. **A** ☐  $100$   
 b. **B** ☒  $100$   
 c. **C** ☐  $100$   
 d. **D** ☐  $100$

# Multiple Choice Stoichiometry Test

**David Wilson**



## Multiple Choice Stoichiometry Test:

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