

Renewable and Nonrenewable Energy Test

Question #1

Fossil fuels are formed

- (A) in shallow areas over short periods of time.
- (B) from the remains of organisms.
- (C) in underground reservoirs that are relatively cool.
- (D) by high winds and temperatures.

Question #2

Which of these resources is renewable?

- (A) natural gas
- (B) oil
- (C) coal
- (D) wind

Question #3

Which process uses living organisms to produce energy?

- (A) oil
- (B) gas
- (C) coal
- (D) solar

Question #4

Which renewable energy source is too much electricity is thought to be available?

- (A) oil
- (B) coal
- (C) wind
- (D) nuclear

Question #5

Which is a source of renewable energy?

- (A) oil
- (B) hydropower
- (C) natural gas
- (D) nuclear weapon

Question #6

Which is a nonrenewable resource?

- (A) oil
- (B) trees
- (C) solar energy
- (D) fossil crops

Question #7

What is responsible for changing the remains of prehistoric sea animals into natural gas?

- (A) time
- (B) water
- (C) gravity
- (D) pressure

Question #8

Which condition does not affect energy?

- (A) time
- (B) coal
- (C) oil
- (D) solar

Question #9

Which number indicates the use of nonrenewable resources?

- (A) shiny rocks
- (B) shiny stones
- (C) shiny sand
- (D) shiny rocks, stones, and sand that are nonrenewable to produce energy.

Question #10

Which nonrenewable resource is not energy? It is limited resources. Perpetual uses are developing devices that will make renewable energy more available. Which is a source of nonrenewable energy?

- (A) coal
- (B) gasoline
- (C) oil
- (D) wind

Non Renewable Energy Study Guide Answer Key

RC Schank

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

acids and bases definition examples properties uses with - May 08 2022

acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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feb 21 2022 both acids and bases are important enough that we devote an entire chapter to them their properties and their reactions 8 1 brønsted lowry definition of acids and bases 8 2 water is both an acid and a base

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jun 12 2019 acids and bases in aqueous solutions will conduct electricity because they contain dissolved ions therefore acids and bases are electrolytes strong acids and bases will be strong electrolytes weak acids and bases will be weak electrolytes

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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions h when dissolved in water and a base is a substance that forms hydroxide ions oh when dissolved in water

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mar 24 2021 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonumber

14 2 acids properties and examples chemistry libretexts - Apr 19 2023

may 20 2018 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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nov 13 2022 suggest simple tests you could carry out to determine if an unknown substance is an acid or a base state the chemical definitions of an acid and a base in terms of their behavior in water write the formula of the salt formed when a given acid and base are combined

