

**BAGONG SILANG NATIONAL HIGH SCHOOL**  
**1<sup>ST</sup> QUARTER EXAMINATION**  
**PHYSICAL SCIENCE**

NAME: \_\_\_\_\_  
GRADE: \_\_\_\_\_

DATE: \_\_\_\_\_  
SCORE: \_\_\_\_\_

Directions: Write the letter of the correct answer on the space provided.

- \_\_\_\_\_ 1. It is also known as nuclear fusion and the formation of new nuclei actions in the early stages of development of the universe.  
a. Nucleosynthesis                      b. S-Process                      c. R-Process                      d. Singularity
- \_\_\_\_\_ 2-4. There are two categories of star, the average star and massive star. The life cycle of these two stars are common from 1<sup>st</sup> to 3<sup>rd</sup> stages namely nebula, protostar, and main sequence star. What will be the next stages in the life of an average star until it reaches its death?  
a. Red super giant → Supernova → Black hole → neutron star 0  
b. Red giant → white dwarf → black dwarf 3  
c. Red giant → white dwarf → black hole 2  
d. Red supergiant → white dwarf → black dwarf 1
- \_\_\_\_\_ 5. He synthesized element with atomic number 43 using a linear particle accelerator.  
a. Ernest Rutherford      b. Ernest Lawrence      c. Dmitri Mendeleev      d. John Dalton
- \_\_\_\_\_ 6. The following are light elements EXCEPT:  
a. helium                      b. hydrogen                      c. lithium                      d. iron
- \_\_\_\_\_ 7. Which of the following molecule is nonpolar?  
a. NaCl                      b. HCl                      c. CO<sub>2</sub>                      d. NH<sub>3</sub>
- \_\_\_\_\_ 8. What type of chemical bond holds the atoms of water molecule together?  
a. Hydrogen bond      b. Ionic bond                      c. Polar covalent bond      d. Nonpolar covalent bond
- \_\_\_\_\_ 9. Which of the following IMFAs is considered as the weakest?  
a. H-bonding                      b. Ion-dipole                      c. Dipole-dipole                      d. London forces
- \_\_\_\_\_ 10. Which of the following statements is TRUE about nonpolar molecules?  
a. high boiling point                      c. low surface tension  
b. high melting point                      d. low vapor pressure
- \_\_\_\_\_ 11. Forces that are present between an ion and a polar molecule  
a. ion-dipole                      b. dipole-dipole                      c. dispersion forces                      d. hydrogen bonding
- \_\_\_\_\_ 12. H-bonding forms when the substances involved are polar and there are \_\_\_\_\_.  
a. H-atoms of a molecule interact with to O, N, F of other molecules.  
b. C-atoms attached to O, N, F.  
c. central atoms with O, N, F as attached atoms.  
d. unshared pair of electrons in the central atom.
- \_\_\_\_\_ 13. The amount of energy required to stretch or increase the surface of a liquid by a unit area  
a. specific heat                      b. surface tension                      c. vapor pressure                      d. heat of vaporization
- \_\_\_\_\_ 14. Liquids can form spherical elastic film to minimize surface area. What intermolecular force is responsible for the formation of this film in water?  
a. H-bonding                      b. ion-induced dipole                      c. dipole-induced dipole      d. London dispersion force
- \_\_\_\_\_ 15. Which group of biomolecules carries and passes on the hereditary information of the organisms?  
a. carbohydrates                      b. lipids                      c. nucleic acids                      d. proteins
- \_\_\_\_\_ 16. What are the elements that make up carbohydrates?  
a. C and H                      b. C, H and N                      c. C, H and O                      d. C, H, O and N
- \_\_\_\_\_ 17-19. Is carbohydrates one of the most important biological macromolecules needed by a human body?  
a. Yes, because it is the primary source of energy. 3  
b. Yes, because it can be a storage of energy for a long period of time. 1  
c. Yes, because it can be found in rice, cereals, fruits and pastas. 2  
d. Yes, because it carries genetic information. 0
- \_\_\_\_\_ 20. The measure of how fast is the change in the concentration of the reactants or products in a chemical reaction.  
a. Activation Energy                      b. Activation Rate                      c. Chemical Reaction                      d. Concentration
- \_\_\_\_\_ 21. A chemical change that occurs when two or more substances combine to form a new substance.  
a. Activation Energy                      b. Chemical Kinetics                      c. Chemical Reaction                      d. Product

# Section 2review Answer Key Physical Science

**Kaplan Nursing**



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*Admission Test* Lawrence Solomon, Morris Bramson, 1988 Including more material and better coverage than any of the competition this completely revised and enlarged guide incorporates four full length sample tests complete with coverage of the new essay question     *"The" Athenaeum* ,1890     The Bookseller and the Stationery Trades' Journal ,1896 Official organ of the book trade of the United Kingdom     **The Spectator** ,1872     *Athenaeum* ,1874     **CliffsTestPrep CSET: Multiple Subjects** Jerry Bobrow, Stephen Fisher, 2007-05-04 The CliffsTestPrep series offers full length practice exams that simulate the real tests proven test taking strategies to increase your chances at doing well and thorough review exercises to help fill in any knowledge gaps CliffsTestPrep CSET can help you prepare for the California Subject Examination Test Multiple Subjects The Commission on Teacher Credentialing uses the CSET to evaluate subject matter competence for instructors seeking the Multiple Subject Teaching Credential Inside this test prep tool you ll find Full length practice tests with answers and in depth explanations Analysis of exam areas and question types with emphasis on suggested approaches and samples Intensive review of subjects using outlines glossaries and diagnostic tests Introduction to the format and scoring of the exam overall strategies for answering multiple choice questions and questions commonly asked about the CSET Some test taking tips and reminders to put candidates on the right track This book will help you understand the types of questions that will test your knowledge in seven general areas including Visual and Performing Arts You can get ready to show what you know in topics such as Sentence structure preferred usage and conventional forms of spelling capitalization and punctuation in written English United States and California history of early exploration through modern day economic political and cultural development The fundamentals of mathematics with focus on prime numbers factors integers ratio area volume perpendicular and more Primary scientific concepts principles and interrelationships in the context of real life problems and significant science phenomena and issues Concepts of biomechanics that affect movement and the critical elements of basic movement skills Social development of children and young adolescents including persons with special needs Components of dance music theatre and visuals arts education With guidance from the CliffsTestPrep series you ll feel at home in any standardized test environment For additional help be sure to visit the Test Prep Think Tank for free online resources

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