

1. What is a large circular mass that revolves around a star?

- |                                     |                                      |
|-------------------------------------|--------------------------------------|
| <input type="radio"/> A Planet      | <input type="radio"/> B Solar System |
| <input type="radio"/> C Spring Tide | <input type="radio"/> D Neap Tide    |

2. What is a system of objects that revolve around a star?

- |                                      |                                |
|--------------------------------------|--------------------------------|
| <input type="radio"/> A Gravity      | <input type="radio"/> B Planet |
| <input type="radio"/> C Solar System | <input type="radio"/> D Orbit  |

3. What is a conceptual framework that explains how planetary systems form?

- |                                    |  |
|------------------------------------|--|
| <input type="radio"/> A Gravity    | <input type="radio"/> B Nebular Theory |
| <input type="radio"/> C Revolution | <input type="radio"/> D Solar System   |

4. What is a force that exists between any two objects that have mass?

- |  |                                      |
|--|--------------------------------------|
| <input type="radio"/> A Gravity        | <input type="radio"/> B Solar System |
| <input type="radio"/> C Nebular Theory | <input type="radio"/> D Revolution   |

5. What is a type of tide that occurs when the sun, moon, and Earth are in a straight line? It is also when the alignment and gravity from the moon and the sun pull the water in the same direction?

- |                                   |                                     |
|-----------------------------------|-------------------------------------|
| <input type="radio"/> A Orbit     | <input type="radio"/> B Seasons     |
| <input type="radio"/> C Neap Tide | <input type="radio"/> D Spring Tide |

6. What is produced when the sun, moon, and Earth align at a 90° angle? It is also when the water on Earth is under the least amount of gravitational pull because the sun's and moon's forces are opposing each other, producing the smallest rise and fall of tidal levels.

- |                                 |                                     |
|---------------------------------|-------------------------------------|
| <input type="radio"/> A Seasons | <input type="radio"/> B Spring Tide |
|---------------------------------|-------------------------------------|

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web sep 22 2021 vsepr theory the vsepr valence shell electron pair repulsion model is used to predict the geometry of molecules based on the number of effective electron pairs around a central atom

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web introduction molecular shape and polarity greatly influence the chemical and physical properties of compounds because a molecule will always assume the geometry that gives it the lowest potential energy molecular shape can

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web draw lewis structure use vsepr to determine molecular geometry determine bond polarity based on electronegativity differences determine molecular polarity based on bond dipoles molecular geometry for the following molecules complete this

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web two regions of electron density around a central atom in a molecule form a linear geometry three regions form a trigonal planar geometry four regions form a tetrahedral geometry five regions form a trigonal bipyramidal geometry and six regions form an octahedral geometry

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web molecules composed of covalently bonded atoms may also be polar or nonpolar for the molecule to be polar it must of course have polar bonds but the key factor for determining the polarity of a molecule is its shape if the polar bonds dipoles are symmetrical around the central atom they offset each other and the resulting molecule is

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linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a tetrahedral geometry all angles 109.5 degrees

**molecular geometry and polarity phet tiss** - Mar 11 2023

in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule