

19.2 Hydrogen Ions and Acidity

- The reaction in which water molecules produce ions is called the self-ionization of water
- $\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- $2\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$



19 2 Hydrogen Ions And Acidity

**American Electroplaters' Society.
Convention**

19 2 Hydrogen Ions And Acidity:

Carbonic Acid Compounds and Hydrogen Ion Activities in Blood and Salt Solutions Erik Johan Warburg,1922

Acidic Deposition: Terrestrial, materials, health and visibility effects Patricia M. Irving,1991 *Journal of the American Medical Association* American Medical Association,1924 Includes proceedings of the association papers read at the annual sessions and lists of current medical literature *The Effects of Acidity and Alkalinity on Legumes* Ollie Clifton Bryan,1922

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