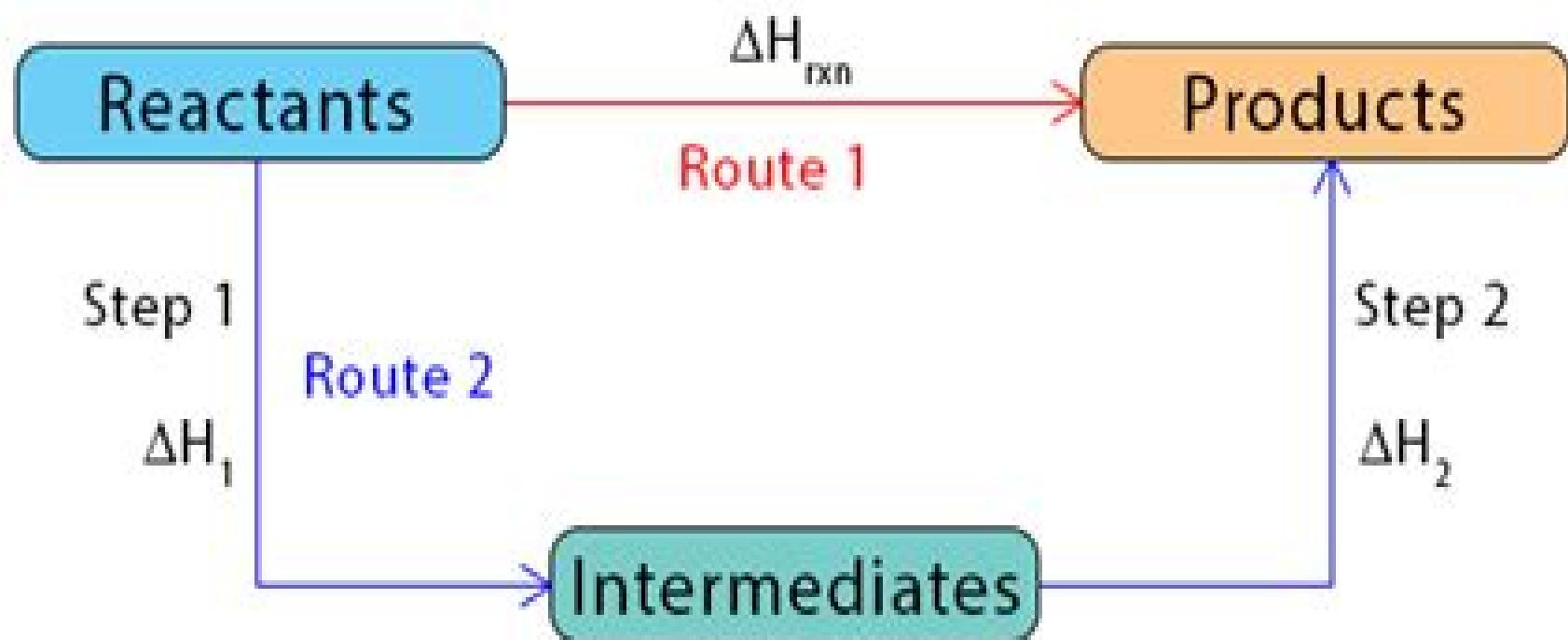


Hess's Law

Total enthalpy change (ΔH) of a reaction equals to the sum of all the enthalpy changes occurring in each step of the reaction



$$\Delta H_{rxn} = \Delta H_1 + \Delta H_2$$

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John Dworetzky

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effects for the region The book discusses the role of Mongolia in the context of the power dynamics in Northeast Asia A relatively small country in terms of its population Mongolia has rarely been examined in this context Sustainable Peace in Northeast Asia makes a fresh assessment of its potential role *Journal of the Experimental Analysis of Behavior* ,1972

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