

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is the most common primary gas used?

1. How many moles of butane gas (C_4H_{10}) is a cylinder at typical conditions? (1000)
a. 1
b. 2
c. 3
d. 4
2. At constant T , P , V , n , what factor determines whether a substance is a gas?
a. C_p
b. C_v
c. C_p/C_v
d. C_v/C_p
3. The universal gas constant is the same for:
a. all of the above
b. gases only
c. liquid only
d. solid only
4. For butane gas, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$.
a. 1.10
b. 1.10
c. 1.10
d. 1.10
5. What is the ratio of the molar specific heats of a diatomic gas at constant pressure to constant volume?
a. 1.40
b. 1.40
c. 1.40
d. 1.40
6. What is the ratio of the molar specific heats of a diatomic gas at constant pressure to constant volume?
a. 1.40
b. 1.40
c. 1.40
d. 1.40
7. A gas cylinder is filled with gas at 100°C . At this temperature, the gas is at the critical temperature of the gas.
a. 100
b. 100
c. 100
d. 100
8. For butane gas, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$.
a. 1.10
b. 1.10
c. 1.10
d. 1.10
9. At constant T , P , V , n , what factor determines whether a substance is a gas?
a. C_p
b. C_v
c. C_p/C_v
d. C_v/C_p
10. The universal gas constant is the same for:
a. all of the above
b. gases only
c. liquid only
d. solid only
11. For butane gas, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$, $\text{C}_p/\text{C}_v = 1.10$, $\text{C}_p = 1.10 \text{ J/K mol}$, $\text{C}_v = 1.10 \text{ J/K mol}$.
a. 1.10
b. 1.10
c. 1.10
d. 1.10

Multiple Choice Stoichiometry Test

John T. Moore, Richard H. Langley



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