

ACS Final Exam Practice 1

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is $3.0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$. What is this speed in $\text{nm}\cdot\text{ms}^{-1}$?

- (A) $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$ (B) $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$ (C) $3.0 \times 10^{14} \text{ nm}\cdot\text{ms}^{-1}$ (D) $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

- (A) 1.3 g (B) 1.30 g (C) 1.298 g (D) 1.2980 g

PQ-5. What is the name of $\text{Ti}_2(\text{PO}_4)_3$?

- (A) titanium phosphate (B) titanium(III) phosphate
(C) titanium(IV) phosphate (D) titanium tetraphosphate

PQ-6. The formula of strontium hexafluorosilicate is SrSiF_6 . What is the formula of aluminum hexafluorosilicate?

- (A) AlSiF_6 (B) $\text{Al}_2(\text{SiF}_6)_3$ (C) Al_2SiF_6 (D) $\text{Al}_2(\text{SiF}_6)_2$

PQ-8. What is the correct name for N_2O_5 ?

- (A) nitrogen(II) oxide (B) nitrogen(V) oxide
(C) dinitrogen oxide (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of $12.0 \text{ g}\cdot\text{cm}^{-3}$, what is the mass of a nanoparticle with a volume of $1.84 \times 10^{-23} \text{ cm}^3$?

- (A) $2.21 \times 10^{-23} \text{ g}$ (B) $1.63 \times 10^{-20} \text{ g}$ (C) $6.52 \times 10^{-21} \text{ g}$ (D) $1.53 \times 10^{-22} \text{ g}$

70 Question Chemistry Final Exam Answers

G. S. Turpin



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Encyclopedia of Chemical Processing and Design John J. McKetta Jr,1993-06-02 Written by engineers for engineers with over 150 International Editorial Advisory Board members this highly lauded resource provides up to the minute information on the chemical processes methods practices products and standards in the chemical and related industries

Lessons in Organic Chemistry G. S. Turpin,1894 **Inorganic Chemistry for Beginners** Henry Enfield Roscoe,Joseph Lunt,1897 **The Publishers Weekly** ,1892 *Science Reporter* ,1994 **Boston Journal of Chemistry and Popular Science Review** ,1881 Report Upon the Survey of the University of Wisconsin Wisconsin. State Board of Public Affairs,1914 A Class-book of Inorganic Chemistry; with tables of chemical analysis, etc David MORRIS (B.A.),1870

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